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# Stoichiometry Stumper 1 Answer Forensic

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Stoichiometry Stumper 1 Answer Forensic

Stoichiometry Stumper 1 Answer

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stoichiometric calculations  
practice page***

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**Systematic  
Forensics** Stoichiometry  
Stumper 1 Answer  
Forensic stoichiometry  
stumper #1 So, Suzanne  
did not have enough  
benzene to create the  
amount of ethyl benzene  
she claimed to have  
made. But she did have  
enough benzene to create  
the same amount of di-  
chloro benzene that killed  
her  
husband..stoichiometry  
stumper #1 by Amy Hinds  
- Prezi Transcribed Image

Text Stoichiometry  
Stumper #1 You are a  
forensic scientist. You are  
investigating a murder  
involving poison. The  
victim was poisoned with  
a compound called di-  
chloro benzene whose  
formula is  
 $C_6H_4Cl_2$ . Solved:  
Stoichiometry Stumper  
#1 You Are A Forensic  
Scientist ... WML Held 12-03  
Stoichiometry Stumper  
#1 You are a forensic  
scientist. You are  
investigating a murder  
involving  
poison. Stoichiometry  
Stumper #1 Stoichiometry  
Stumper 1 Answer  
Forensic Eventually, you  
will definitely discover a  
additional experience and  
endowment by spending  
more cash. nevertheless  
when? complete you  
receive that you require  
to get those all needs with  
having significantly  
cash? Stoichiometry  
Stumper 1 Answer  
Forensic Stoichiometry  
Stumper #1 You are a  
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involving poison. The  
victim was poisoned with  
a compound called di-  
chloro benzene whose  
formula is  
 $C_6H_4Cl_2$ . Stoich\_Stumpers  
- Stoichiometry  
Stumper #1 You are a  
...Name: \_\_\_\_\_ WML Held  
12-03 Stoichiometry

Stumper #1 You are a forensic scientist. You are investigating a murder involving poison. The victim was poisoned with a compound called dichloro benzene whose formula is  $C_6H_4Cl_2$ . Autopsy results show that the victim's body contained about 22 g of the poison, but the actual amount could have been slightly higher due to tissue absorption.5 - Stoichiometry Stumpers (pick 2 write 1) - Name ...\$1.25 Monday, February 17, 2014 Procedure: Is she lying? 1. Convert 15 grams into moles by dividing by its mass, 78.114 g  $C_6H_6$ . 2. Multiply the moles by the mass of  $C_6H_5CH_3$ . 3. You will get an answer of 18 g. Math Three forensic scientists are investigating a murder involvingStoichiometry Stumper by Princess Bekah - PreziStoichiometry Stumper 1 Answer Forensic Here are 305 of the best book subscription services available now. Get what you really want and subscribe to one or all thirty. You do your need to get free book access. Introduction to Forensic Chemistry// Module 1 Forensic Toxicology Lab Medical Detectives (Forensic Files) - Season

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 #1 You are a forensic  
 scientist. You are  
 investigating a murder  
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 Name: \_\_\_\_\_ WML Held  
 12-03 Stoichiometry  
 Stumper #1 You are a  
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 Autopsy results show that  
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 the poison, but the actual  
 amount could have been  
 slightly higher due to  
 tissue absorption.

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 scientist. You are  
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 victim was poisoned with  
 a compound called di-  
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Stoichiometry Stumper #1 You are a forensic scientist. You are investigating a murder involving poison. The victim was poisoned with a compound called dichloro benzene whose formula is  $C_6H_4Cl_2$ .

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check purity. However most commonly it is used for mixing/making up your solutions used.

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Mass-to-Mass (3 Step)  
Stoichiometry Problem Solving 1. Determine the mass of lithium hydroxide produced when .38g of lithium nitride reacts with water.  $\text{Li}_3\text{N} + 3\text{H}_2\text{O} \rightarrow \text{NH}_3 + 3\text{LiOH}$   
2. Find the mass of sugar ( $\text{C}_6\text{H}_{12}\text{O}_6$ ) required to produce 1.82g of carbon dioxide gas.  $\text{C}_6\text{H}_{12}\text{O}_6 + 2\text{C}_2\text{H}_6\text{O} \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$

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Stoichiometry Stumper

#1

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So, Suzanne did not have

enough benzene to create

the amount of ethyl

benzene she claimed to

have made. But she did

have enough benzene to

create the same amount

of di-chloro benzene that

killed her husband..

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