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Structures Ionic and Covalent Bonds Made Easy Chemical Bonding - Ionic vs.
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 a covalent bond formed
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 force that occurs when a
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 is also weak bonded to an
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the bonding electrons are
pulled closer to one atom,
and the bond is polar.
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Covalent Bonding Single
Covalent Bonds When
only one pair of electrons
is shared, such as in a
hydrogen molecule, it is a
singleChapter 8 Covalent
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Covalent Bonding and
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2 molecule. More
sophisticated descriptions
of chemical bonding will
be discussed in Chapter 9.

8.3 Lewis Structures OWL
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 Lewis structure. Chapter 8:
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covalent, Above 1.7 is an

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Ferromagnetism These

atoms have unpaired

electrons when exposed

to a magnetic field, the

electron spin and line up

parallel

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bonding electrons are

pulled closer to one atom,
and the bond is polar.

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Covalent Bonding Single

Covalent Bonds When

only one pair of electrons

is shared, such as in a

hydrogen molecule, it is a

single

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