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\_\_\_\_ PHASES OF MATTER Date \_\_\_\_

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Law If a balloon is pulled over the neck  
of a flask, and the setup is placed on a

hot plate, the balloon blows up as it  
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2 Answer the questions below based on  
the preceding examples. 1. How many  
times faster does hydrogen sulfide

effuse than krypton? 2. What is the molar mass of a gas that effuses 2.00 times faster than methane (CH<sub>4</sub>)? 3.C:UsersEvan P. SilbersteinDocumentsChemistryFrischChem ...The Ideal Gas - Evan's Chemistry Corner Chemistry: Form WS6.5.1A ... Answer the questions below based on your reading above and on your knowledge of chemistry. ... 4. The relationship between the Kelvin temperature and volume of a gas is linear until the temperature begins to ... WS6-5-1A\_The\_Ideal\_Gas.pdfchemistry form ws6 4 1a answers - Free PDF Documents SharingChemistry: Form WS6.5.1A Name \_\_\_\_ PHASES OF MATTER Date \_\_\_\_ Period \_\_\_\_ The Ideal Gas How do gases behave? In order to study gases, chemists have devised a

model. The model is called an ideal gas (a gas which explains the behavior of all gases).The Ideal Gas - Evan's Regents Chemistry CornerChemistry: Form WS6.3.1A Name \_\_\_\_ PHASES OF MATTER Date \_\_\_\_ Period \_\_\_\_ Phase Changes Heating a substance in a given phase generally causes the temperature to increase. This causes the particles to move faster and collide harder. Harder,C:Documents and SettingsEvanDesktopChemistryChemistry (New ...chemistry form ws6 4 1a answers.pdf FREE PDF DOWNLOAD NOW!!! Source #2: chemistry form ws6 4 1a answers.pdf FREE PDF DOWNLOAD Periodic Table of Elements and Chemistry www.chemcool.com "The chemistry of an atom depends only on the number of electrons, which equals

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the questions below based on the preceding examples. 1. How many times faster does hydrogen sulfide effuse than krypton? 2. What is the molar mass of a gas that effuses 2.00 times faster than methane (CH<sub>4</sub>)? 3.

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Phases of matter change by gaining or losing energy.. Heat decides the state of matter. Adding heat or taking off heat brings in change of state/ phase. Solid --- --> Liquid This is called melting process (solid needs to be heated, example melting of butter, solid on taking up heat changes to liquid)

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