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ANSWER: $KE = \frac{1}{2}mv^2$ so solving for the velocity $V = \sqrt{2KE/m} = 484 \text{ m/sec}$ since substitution yields $m = 5.31 \times 10^{-26}$; $KE = 6.21 \times 10^{-21}$; $V = \sqrt{2 * KE / m} = 483.63 \text{ m/sec}$

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QUESTION: At room temperature, an oxygen molecule with a mass of 5.31×10^{-26} kg typically has a kinetic energy KE of about 6.21×10^{-21} J . How fast is the oxygen molecule moving?

ANSWER: $KE = \frac{1}{2}mv^2$
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 $v = \sqrt{2KE/m} = 484$ m/sec
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 $m = 5.31 \times 10^{-26}$; $KE = 6.21 \times 10^{-21}$; $v = \sqrt{2 * KE / m} = 483.63$ m Problem #16

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• The centripetal force may be provided by friction, gravity, tension, the normal force, or others. •
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