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lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that atoms are conserved, together with knowledge of the chemical properties of the elements involved, can be used to describe and predict chemical reactions.

Tenth grade Lesson Gizmo: Modeling Chemical Reactions

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Using the chemical equations in the Gizmo as a guide, match the following definitions to the type of reaction.

b___ one reactant is broken down into two or more products.

_a___ Two or more reactants combine to form one product.

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Balancing Chemical Equations Answer Key Vocabulary: coefficient, combination, compound, decomposition, double replacement, element, molecule, product, reactant, single replacement, subscript Prior Knowledge Questions (Do these BEFORE using the Gizmo.) [Note: The purpose of these questions is to activate prior knowledge and get students thinking.

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This is the fourth lesson in the chemical reaction unit that covers HS-PS1-7, the uses of mathematical representations to support the claim that atoms, and therefore mass, are conserved during a chemical reaction; and DCI-PS1.B, the fact that atoms are conserved, together with knowledge of the chemical

properties of the elements involved, can be used to describe and predict chemical reactions.

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The law of conservation of matter states that no atoms are created or destroyed in a chemical reaction. Therefore, a balanced chemical equation will show the same number of each type of atom on each side of the equation. To set up an equation in the Chemical Equations Gizmo™, type the chemical formulas into the text boxes of the

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