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Explain your answer. 10, A scuba tank contains a mixture of

oxygen and helium gases, Before the dive, the partial pressure of

oxygen gas is 0.65 atm and the partial pressure of helium is 0.38

atn. a. What is the total pressure in the scuba tank before the

dive? b. After the dive, the partial pressure due to the oxygen gas

is reduced by 80%.

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186 POGIL™ Activities for High School Chemistry 1. In Model 1,

what does a dot represent? ... Which gas variable (P internal, V, T

or n) is most closely related to the length of the arrows in Model

1? ... In your answer, consider the external and internal pressure

data given in Model 1.

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