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# Chapter 15 Water And Aqueous Systems Test Answers

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Chapter 15 Water And Aqueous

SECTION 15.1 WATER AND ITS PROPERTIES  
(pages 445-449)

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| SYSTEMS         | Systems -      | students learn  |
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| Chemistry .     | 507: 21        | associated      |
| More Chapters   | Answer         | with water      |
| Of Chemistry    | Colloids       | and aqueous     |
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| Chemistry       | separated      | ce Hall         |
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| Chemistry 121   | because their  | Water and       |
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| Review .        | those of       | activities to   |
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| Quizzes. What   | and they do    | the             |
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| <p>Chapter 15 "Water and Aqueous Systems" Chapter 15 (Water and Aqueous Systems) Test Study Guide □ The bonds between the hydrogen and oxygen atoms in a water molecule are polar covalent bonds. □ The covalent bonds are polar because the oxygen atom has a greater electronegativity than the hydrogen atoms. □ The bonds between adjacent water molecules are called</p> | <p>hydrogen bonds. Chapter 15 (Water and Aqueous Systems) Test - Chapter 15 ... Chapter 15 - Water and Aqueous Systems - 15.1 Water and Its Properties - 15.1 Lesson Check - Page 493: 1 Answer The hydrogen bonding in water cause the high surface tension and low vapor pressure of water. Chemistry (12th Edition) Chapter 15 - Water and Aqueous ... The high surface</p> | <p>tension of water is due to the: a. small size of water molecules. b. low mass of water molecules. c. hydrogen bonding between water molecules. d. covalent bonds in water molecules. ____ 12. Salts and other compounds that remove moisture from air are said to be: a. efflorescent. c. colloidal. b. surfactant. d. hygroscopic. 10 9 ... 05 CTR ch15 7/12/04 8:14 AM Page 387 WATER</p> |
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| <p>AND<br/>AQUEOUS<br/>...Test and<br/>improve your<br/>knowledge of<br/>Prentice Hall<br/>Chemistry<br/>Chapter 15:<br/>Water and<br/>Aqueous<br/>Systems with<br/>fun multiple<br/>choice exams<br/>you can take<br/>online with<br/>Study.comPre<br/>ntice Hall<br/>Chemistry<br/>Chapter 15:<br/>Water and<br/>Aqueous<br/>...CHAPTER 15<br/>  Aqueous<br/>Equilibria:<br/>Chemistry of<br/>the Water<br/>World 15.1.<br/>Collect and<br/>Organize<br/>Figure<br/>P15shows .1<br/>four lines to</p> | <p>describe the<br/>possible<br/>dependence<br/>of percent<br/>ionization of<br/>acetic acid<br/>with<br/>concentration.<br/>We are to<br/>choose the<br/>one that best<br/>represents the<br/>trend for this<br/>weak<br/>acid.CHAPTER<br/>15   Aqueous<br/>Equilibria:<br/>Chemistry of<br/>the Water<br/>WorldThe<br/>Water<br/>Molecule: a<br/>Review Water<br/>is a simple tri-<br/>atomic<br/>molecule, H 2<br/>O Each O-H<br/>bond is highly<br/>polar, because<br/>of the high<br/>electronegativ<br/>ity of the</p> | <p>oxygen (N, O,<br/>F, and Cl have<br/>high values)<br/>bond angle of<br/>water = 105o<br/>due to the<br/>bent shape,<br/>the O-H bond<br/>polarities do<br/>not cancel.<br/>This means:<br/>water is a<br/>polar<br/>molecule.<br/>Chapter 15<br/>Water And<br/>Aqueous<br/><i>Chapter 15:<br/>Water and<br/>Aqueous<br/>Systems<br/>Flashcards  <br/>Quizlet</i><br/>daily from<br/>chapter 15<br/>water and<br/>aqueous<br/>systems<br/>worksheet<br/>answers,<br/>source:trajans<br/>cimed.com</p> |
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### **Chapter 15 Water And Aqueous**

The Water and Aqueous Systems chapter of this Prentice Hall Chemistry Companion Course helps students learn the essential

lessons associated with water and aqueous systems.

#### *SECTION 15.1 WATER AND ITS PROPERTIES (pages 445-449)*

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The Water

Molecule: a Review Water is a simple tri-atomic molecule, H<sub>2</sub>O Each O-H bond is highly polar, because of the high electronegativity of the oxygen (N, O, F, and Cl have high values) bond angle of water = 105° due to the bent shape, the O-H bond polarities do not cancel. This means: water is a polar molecule.  
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| <p>activities to<br/>yourself study<br/>the<br/>vocabulary<br/>and major<br/>concepts<br/>presented in<br/>this chapter<br/><i>CHAPTER 15  <br/>Aqueous<br/>Equilibria:<br/>Chemistry of<br/>the Water<br/>World</i><br/>Chapter 15-<br/>Water and<br/>Aqueous<br/>Systems.<br/>STUDY.<br/>Flashcards.<br/>Learn. Write.<br/>Spell. Test.<br/>PLAY. Match.<br/>Gravity.<br/>Created by.<br/>Joseph_Wang6<br/>. Terms in this<br/>set (233)<br/>Water is an<br/>amazing liquid<br/>that makes.<br/>plant and</p> | <p>animal life on<br/>Earth Possible.<br/>In addition to<br/>water forming<br/>the Earth's<br/>oceans water,<br/>it also forms<br/>the.<br/><u>Chapter 15-<br/>Water and<br/>Aqueous<br/>Systems</u><br/><u>Flashcards  <br/>Quizlet</u><br/>Water And<br/>Aqueous<br/>Systems<br/>Chapter 15<br/>Chemistry . ...<br/>WATER AND<br/>AQUEOUS<br/>SYSTEMS<br/>&amp;nbs p;<br/>chapter 15<br/>Chemistry .<br/>More Chapters<br/>Of Chemistry<br/>Quizzes.<br/>Chemistry<br/>Chapter 1<br/>Quiz (part 1)<br/>Chemistry 121</p> | <p>Chapter 1 Part<br/>2 . Chemistry<br/>Review .<br/>Featured<br/>Quizzes. What<br/>Animal Would<br/>Your Daemon<br/>Settle As?<br/><u>Chapter 15<br/>water and<br/>aqueous<br/>systems -<br/>SlideShare</u><br/>Start studying<br/>Chapter 15-<br/>Water and<br/>Aqueous<br/>Systems.<br/>Learn<br/>vocabulary,<br/>terms, and<br/>more with<br/>flashcards,<br/>games, and<br/>other study<br/>tools.<br/><i>Chapter 15<br/>Water and<br/>Aqueous<br/>Systems<br/>Worksheet<br/>Answers ...</i></p> |
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| Chapter 13 - States of Matter  | 159 SECTION 15.1   | <b>Chapter 15 (Water and Aqueous Systems) Test - Chapter 15 ...</b>                       |
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| Chapter 16 - Solutions   |  | Check - Page 493: 1 Answer  |
| Chapter 17 - Thermochemistry Chapter 18 - Reaction Rates and Equilibrium   |  | The hydrogen bonding in water cause the high surface tension and                          |
| Chapter 19 -   | affects the  | low vapor   |

pressure of water.  
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WATER AND  
AQUEOUS ...  
The high surface tension of water is due to the: a. small size of water molecules. b. low mass of water molecules. c. hydrogen bonding between water molecules. d. covalent bonds in water molecules.  
\_\_\_\_ 12. Salts and other compounds that remove moisture from air are said to

be: a. efflorescent. c. colloidal. b. surfactant. d. hygroscopic.  
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Water and  
Aqueous ...  
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Systems) Test  
Study Guide □  
The bonds  
between the  
hydrogen and  
oxygen atoms  
in a water  
molecule are  
polar covalent  
bonds. □ The  
covalent  
bonds are  
polar because  
the oxygen  
atom has a  
greater  
electronegativ  
ity than the

hydrogen atoms. □ The bonds between adjacent water molecules are called hydrogen bonds.  
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**Chapter 15:**  
**Water and**  
**Aqueous**  
**Systems ...**  
CHAPTER 15 |  
Aqueous  
Equilibria:  
Chemistry of  
the Water  
World 15.1.  
Collect and  
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Figure  
P15shows .1  
four lines to  
describe the  
dependence  
of percent  
ionization of  
acetic acid

with concentration. We are to choose the one that best represents the trend for this weak acid.

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1. Chapter 15 "Water and Aqueous Systems" Pre-AP Chemistry Charles Page High School Stephen L. Cotton 2. Section 15.1 Water and its Properties

OBJECTIVES:  
-Explain the high surface tension and low vapor pressure of water in terms of the structure of the water molecule and hydrogen bonding. 3.

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Chapter 15 - Water and Aqueous Systems - 15.3 Heterogeneous Aqueous Systems - 15.3 Lesson Check - Page 507: 21

Answer

Colloids cannot be separated through filtering because their particles are smaller than those of suspensions and they do not settle out over time.

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