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QUALITATIVE ANALYSIS. (PART II) (1) Define qualitative analysis. (2) What metals impart the following colors to the borax bead: (a) green, (b) blue, and (c) amethyst in the oxidizing flame, and colorless in the reducing flame? (3) What is aqua regia, and how is it made? (4) What are the two methods of qualitative analysis? (5) Express, in the form of an equation, the reaction that takes place when silver nitrate is precipitated by hydrochloric acid. (6) Describe the separation of the metals of the third group when phosphoric and oxalic acids are absent. (7) What metals are precipitated by hydrochloric acid? (8) What part of the flame of a Bunsen burner acts (a) as an oxidizing flame? (b) as a reducing flame? (9) How are (a) oxidation and (b) reduction accomplished by the flame of a Bunsen burner? (10) What metal imparts a

crimson color to the flame? (11) What metals give black precipitates when their solutions are treated with hydrogen sulphide? (12) Describe the method of determining the members of the seventh group. For notice of the copyright, see page immediately following the title page. (13) What metal, when fused with sodium carbonate and potassium nitrate, imparts a deep-green color to the fusion? (14) Name the metals that are not precipitated from acid...

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